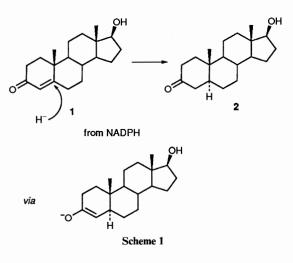
New Routes to 4-Substituted Steroids: Synthesis of 4-Cyanoprogesterone, a Potent Inhibitor of the Enzyme 5α -Reductase

Margret Haase-Held, Maria Hatzis and John Mann*

Department of Chemistry, Reading University, Whiteknights, Reading RG6 2AD, UK

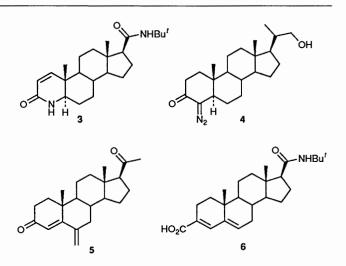
Two new synthetic routes to 4-substituted derivatives of progesterone are described: one involves the palladium-catalysed addition of alkenes to 4- (trifluoromethanesulfonyloxy) progesterone, and the other involves the addition of the anion of acetonitrile to the enol acetate, enol lactone 20-acetoxy-4-oxa-3,5-secopregna-5,17-dien-3-one. This second method provides gram quantities of 4-cyanoprogesterone, which proved to be a potent inhibitor of the enzyme 5α -reductase, and thus of considerable interest as a potential agent for the treatment of benign prostatic hyperplasia and prostatic cancer.

The enzyme 5α -reductase catalyses the conversion of the major male hormone testosterone 1 into dihydrotestosterone 2 via the mechanism shown in Scheme 1.¹ This reduction is a key reaction in androgen metabolism since the relative levels of testosterone and dihydrotestosterone are important for the control of, *inter alia*, the male sex drive and the size of the prostate. In particular, increased levels of dihydrotestosterone are implicated in the induction of acne, recession of scalp hair and prostate enlargement in later life. A large percentage of males over the age of 50 years suffer from this last affliction (benign prostatic hyperplasia), and a proportion of these (around 11 000 new cases each year in the UK) eventually suffer from prostatic cancer.



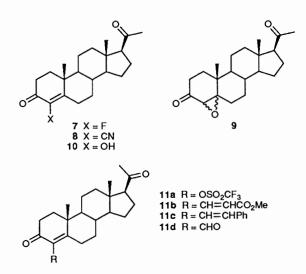
Clearly inhibition of 5α -reductase with a consequent reduction in the production of dihydrotestosterone could represent a viable strategy for the control of prostatic hyperplasia and prostatic cancer, and a number of inhibitors have already been developed. Of these, compounds 3 to 6^{2-5} are the most potent, and compound 3 (K_i 25 nmol dm⁻³) has recently been introduced as a prescription drug for the treatment of prostatic hyperplasia in the UK and elsewhere.

Our involvement in this area arose from work on inhibitors of the enzyme aromatase (for the treatment of breast cancer),⁶ and the discovery that 4-fluoroprogesterone 7 was an inhibitor of both aromatase and 5α -reductase at the micromolar level. It appeared reasonable that the electron-withdrawing substituent at C-4 was stabilising the enolic intermediate shown in Scheme 1 and that the compound was thus acting as a competitive inhibitor of 5α -reductase. We decided to prepare other derivatives of progesterone with electron-withdrawing groups



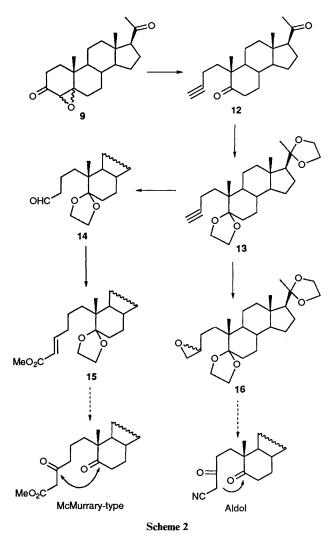
at C-4, and the results of these efforts are the subject of this paper (a preliminary account of this work has been published).⁷

At the commencement of the work, a literature survey revealed one report of a synthesis of 4-cyanoprogesterone⁸ 8, but the synthesis appeared to be rather inefficient and certainly not flexible enough to be used for the synthesis of other 4substituted steroids. Our initial attempt to introduce a substituent at C-4 was based upon the work of Piers and Fleming⁹ involving palladium-catalysed reactions of cyanide with enol trifluoromethanesulfonates (triflates). Reaction of the epoxide mixture 9 with formic acid^{6a} provided 4-hydroxypro-



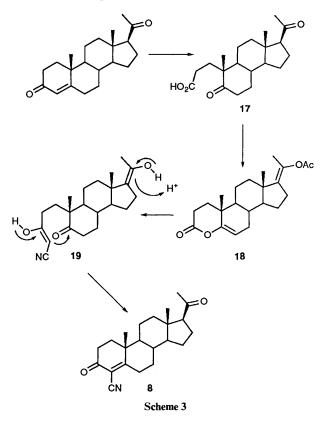
gesterone 10 in poor yield (ca. 40%), although the triflate 11a could be produced in 70% yield using trifluoromethanesulfonic anhydride and pyridine in dichloromethane. Reaction of this with LiCN in the presence of tetrakis(triphenylphosphine)palladium(0) and 12-crown-4 (typically in the ratio of 2:5:0.2:0.3) in dry benzene at room temperature led to complete consumption of the triflate and production of an intractable mixture of products. However, reaction of the triflate 11a with palladium(II) acetate and alkenes (methyl acrylate and styrene) in the presence of triethylamine yielded the 4substituted steroids 11b and 11c in reasonable yields (46 and 86%, respectively). Ozonolysis of 11c then provided 4-formylprogesterone 11d (58%), and this methodology may have general utility for the production of 4-substituted steroids, although not the type of compound we were attempting to prepare.

Our second potential route to 4-substituted derivatives of progesterone is shown in Scheme 2, and involved an initial



cleavage of the A-ring of epoxide 9 using the Eschenmoser method ¹⁰ to produce the keto alkyne 12 (85% isolated yield on the 10 g scale). This was converted into the diketal 13 prior to hydroboration with disiamylborane and production of the diketal aldehyde 14 (52%). Treatment of this with (methoxycarbonylmethylene)triphenylphosphorane yielded the expected unsaturated ester 15 (50% yield). Alternatively, partial hydrogenation of the alkyne 12 (Lindlar catalyst, 95%) followed by treatment with *m*-chloroperbenzoic acid yielded the epoxide 16 (35% yield). The low yields of many of these conversions discouraged us from attempting further reactions in the planned sequence (Scheme 2).

Our third route proved successful and is shown in Scheme 3.



Ozonolysis of progesterone followed by oxidative work-up (H₂O₂-NaOH) produced the diketo acid 17 (80-90% yield on 10 g scale), and this was converted into the enol acetate, enol lactone 18 using a mixture of acetic anhydride and acetyl chloride (ca. 50% on the 5 g scale). Finally, addition of 2 equiv. of the anion from acetonitrile [BuLi, tetrahydrofuran (THF), -78 °C], followed by addition of potassium tert-butoxide and tert-butyl alcohol yielded the desired 4-cyanoprogesterone 8 (40-45%) on the 5 g scale). A proposed mechanism for this interesting conversion is shown in Scheme 3, and it is particularly interesting to note that there was no evidence for the production of steroids with the α -stereochemistry at C-17. This presumably implies that protonation in the conversion of enol 19 into ketone 8 by tert-butyl alcohol occurs exclusively from the α -face. The overall three-step reaction sequence has been carried out routinely on the 5-10 g scale, and although there is clearly room for optimisation (overall yield ca. 20%), the process is operationally very simple.

The availability of gram quantities of 4-cyanoprogesterone meant that comprehensive biological evaluation could be undertaken (at the Cancer Research Campaign Laboratories at Sutton, Surrey), and these experiments will be reported in full in a separate paper. However, it is worth noting that both *in vitro* and *in vivo* compound **8** exhibited marked inhibitory activity against the enzyme 5α -reductase (0.05 and 0.5 µmol, respectively), and it is clearly worth preparing other compounds of this type for biological evaluation. This work is underway.

Experimental

IR spectra were recorded on a Perkin-Elmer 881 double beam grating spectrometer using sodium chloride cells and chloroform solutions unless otherwise stated. ¹H NMR spectra were recorded at 220 MHz using a Perkin-Elmer R34 instrument, and at 400 MHz via the SERC NMR Service at the University of Warwick in deuteriochloroform solution using tetramethylsilane as internal standard. ¹³C NMR spectra were recorded using a JEOL FX90Q spectrometer. J Values are quoted in Hz. M.p.s were obtained using an Electrothermal digital melting point apparatus and are uncorrected.

Reactions requiring anhydrous conditions were carried out under a static nitrogen atmosphere using oven dried glassware. THF was dried over molecular sieves and distilled from sodium-benzophenone prior to use. All other reagents and solvents were used as supplied without further purification. Flash chromatography was performed using SorbosilTM C60 silica gel. Light petroleum refers to the fraction with b.p. 40– 60 °C unless otherwise stated.

 $4(\alpha,\beta),5$ -Epoxypregnane-3,20-dione 9.—Hydrogen peroxide $(30\%; 100 \text{ cm}^3)$ and aq. sodium hydroxide (4 mol dm⁻³; 10 cm³) were added dropwise and at the same time to a solution of progesterone (25.00 g, 0.08 mol) in ethanol (500 cm³) at 0 °C, whilst maintaining the temperature at 0 °C and the pH between 8 and 9. After ca. 7 h further aq. NaOH (20 cm³) was added. The reaction mixture was stored at 0 °C for 3 days, subsequently poured into iced water (1 dm³), and the resulting white precipitate extracted with dichloromethane. The organic extract was evaporated and recrystallisation of the resulting solid from acetone-hexane yielded a mixture of the two title epoxides (22.84 g, 87%; α : β , 1:3); m.p. (α , β -epimers) 130–132 °C; v_{max}/cm⁻¹ 1692 (C=O) and 856 (C-O); $\delta_{\rm H}$ (400 MHz; CDCl₃) 0.61 and 0.62, 1.03 and 1.12, 2.08 and 2.09 (each 3 H, s, 18-, 19- and 21-CH₃ of α , β -epimers, respectively) and 2.95 and 3.00 (1 H, s, 4-H of β , α -epimers, respectively); $\delta_C(22.49 \text{ MHz};$ CDCl₃) 209.16, 209.05 (C-20), 206.61, 206.55 (C-3), 70.03, 69.82 (C-5), 63.42, 63.25 (C-4), 62.66 and 62.47 (C-17) (Found: C, 76.4; H, 9.2. C₂₁H₃₀O₃ requires C, 76.33; H, 9.15%).

4-Hydroxyprogesterone 10.—A solution of progesterone epoxide mixture 9 (19.16 g, 58 mmol) dissolved in formic acid (90%; 170 cm³) was heated under reflux for 1.25 h. The reaction mixture was poured into hot water (800 cm³) and then left to cool to room temperature. The precipitate was filtered off, washed with water and then dissolved in dichloromethane (300 cm³). The organic layer was washed with brine (2 × 100 cm³) and saturated aq. sodium hydrogen carbonate (2 × 100 cm³), dried (MgSO₄) and then evaporated. The residue was dissolved in a little dichloromethane and crystallised from methanol (250 cm³) to give the title compound 10 as colourless crystals; m.p. 205– 215 °C (lit.,⁸ 221–227 °C); $\delta_{\rm H}$ (220 MHz, CDCl₃) 0.64, 1.14 and 2.10 (each 3 H, s, 18-, 19- and 21-CH₃, respectively), 2.46–2.62 (3 H, m), 2.92–3.10 (1 H, m) and 6.08 (1 H, s, OH exchangeable).

4-(Trifluoromethylsulfonyloxy)progesterone 11a.—A solution of 10 (4.87 g, 14.7 mmol) and pyridine (2 cm³, 25 mmol) in dry dichloromethane (70 cm³) was added to a solution of trifluoromethanesulfonic anhydride (2.5 cm³, 15 mmol) in dry dichloromethane (50 cm³) at 0 °C. The mixture was stirred for 2 h at 0 °C and then poured into water (100 cm³). The organic layer was separated, washed with water $(2 \times 250 \text{ cm}^3)$, dried (MgSO₄) and then evaporated. The residue was purified by filtration on silica gel with dichloromethane-ethyl acetate (5:1) as eluent ($R_f 0.74$, SiO₂, dichloromethane-ethyl acetate, 5:1). Crystallisation from diethyl ether gave the title compound 11a (4.83 g, 71%) as colourless crystals; m.p. 118-119 °C; v_{max}/ cm⁻¹ 2944 (C–H), 1698 (C=O), 1410 and 1142; $\delta_{\rm H}$ (220 MHz; CDCl₃) 0.68, 1.28 and 2.13 (each 3 H, s, 18-, 19- and 21-CH₃, respectively), 2.50-2.62 (3 H, m) and 2.92 (1 H, dt, J 13.2, 3.5, 2-H); δ_c(22.49 MHz; CDCl₃) 208.8 (C-20), 188.4 (C-3), 159.7 (C-4), 139.5 (C-5), 118.7 (CF₃), 31.7 (C-21), 17.6 (C-19) and

13.3(C-18)(Found:C,57.1;H,6.35;S,6.9. $C_{22}H_{29}F_{3}O_{5}S$ requires C, 57.13; H, 6.35; S, 6.93%); *m/e* 480 (M + NH₄⁺, 100%).

4-(2-Methoxycarbonylvinyl)progesterone 11b.—Compound 11a (1.57 g, 3.4 mmol) methyl acrylate (0.35 cm³, 4 mmol), triethylamine (0.9 cm³) and palladium acetate (17.8 mg, 0.08 mmol) in dry dimethylformamide (15 cm³) were stirred at 60 °C for 2 h. Diethyl ether (20 cm³) and water (100 cm³) were added, and the organic layer was separated and the aqueous layer was extracted with diethyl ether. The combined organic layers were washed with water, dried $(MgSO_4)$ and then evaporated to afford crude 11b (1.40 g) as a yellow oil. This was purified by flash chromatography on silica gel with dichloromethane-ethyl acetate 5:1 as eluent (R_f 0.60), SiO₂, dichloromethane-ethyl acetate, 5:1) and crystallised from diethyl ether to give the title compound 11b (0.63 g, 46%) as colourless crystals; m.p. 140-144 °C; v_{max}/cm⁻¹ 2945 (C-H), 1701 (C=O), 1677 (C=C), 1438, 1295 and 1174; $\delta_{\rm H}(220 \text{ MHz}; \text{CDCl}_3)$ 0.68, 1.24 and 2.12 (each 3 H, s, 18-, 19- and 21-CH₃, resectively), 2.40-2.62 (3 H, m), 2.95 (1 H, dt, J 15.4, 3.5, 2-H), 3.76 (3 H, s, OCH₃) and AB-signal (δ_A 7.47, δ_B 6.36, J_{AB} 15.4, 2 H, vinyl H) (Found: C, 75.1; H, 8.6. C₂₅H₃₄O₄ requires C, 75.34; H, 8.60%); m/e 398 (M⁺), 366 $(M^+ - CH_3OH)$, 339 $(M^+ - CO_2CH_3)$ and 131 (100%).

4-(2-Phenylvinyl)progesterone 11c.—A mixture of compound 11a (2.97 g, 6.4 mmol), triethylamine (0.8 cm³, 5.7 mmol), styrene (0.75 cm³, 6.6 mmol), palladium acetate (57.4 mg, 0.26 mmol), triphenylphosphine (31.4 mg, 0.12 mmol) in dry dimethylformamide (30 cm³) was stirred at 60-70 °C for 1.5 h. The reaction mixture was poured into water (50 cm³) and extracted with diethyl ether $(3 \times 75 \text{ cm}^3)$. The combined organic phases were washed with brine $(4 \times 50 \text{ cm}^3)$, dried (MgSO₄) and then evaporated. The residue was purified by flash chromatography on silica gel with diethyl ether-light petroleum (4:1) as eluent $(R_{\rm f}\,0.36)$, SiO₂, diethyl ether-light petroleum, 4:1). Crystallisation from diethyl ether-light petroleum gave the title compound 11c (2.37 g, 86%) as pale yellow crystals; m.p. 150–151 °C; v_{max} cm⁻¹ 3017 (C–H), 1701 (C=O) and 1665 (C=C); $\delta_{\rm H}$ (220 MHz; CDCl₃) 0.67, 1.24 and 2.11 (each 3 H, s, 18-, 19- and 21-CH₃, respectively), 2.42-2.64 (3 H, m), 3.10 (1 H, dt, J 13.2, 3.5, 2-H), AB-signal (δ_A 6.82, δ_B 6.66, J_{AB} 15.4, 2 H, vinyl H) and 7.14-7.52 (5 H, m, Ar-H) (Found: C, 83.4; H, 8.72. C₂₉H₃₆O₂ requires C, 83.61; H, 8.71%); m/e 416 (M⁺) and 91 (100%).

4,5-Secopregn-3-yne-5,20-dione 12.—To a solution of $4(\alpha,\beta)$,-5-epoxypregnane-3,20-dione 9 (10.00 g, 0.03 mol) in absolute ethanol (150 cm³) at 40 °C was added toluene-p-sulfonohydrazide (1.1 equiv., 6.20 g). The solution immediately turned yellow and a precipitate formed. After ca. 2 h the precipitate dissolved and the reaction was allowed to proceed for a further 5 h until the solution turned colourless. The reaction mixture was subsequently treated with diethyl ether and water, the organic layer was concentrated, and then the product was extracted into dichloromethane. This crude product was chromatographed on a silica column with diethyl ether-light petroleum (1:1) as eluent and 4,5-secopregn-3-yne-5,20-dione 12 (7.89 g, 0.026 mol, 85%) was isolated; m.p. 148-150 °C; v_{max}/cm^{-1} 3306 (C=C-H), 2252 (C=C) and 1700 (C=O); $\delta_{\rm H}(220 \text{ MHz}; \text{CDCl}_3) 0.65, 1.05 \text{ and } 2.09 \text{ (each 3 H, s, 18-, 19-)}$ and 21-CH₃, respectively) and 2.51 (1 H, m, terminal C-H); $\delta_{\rm C}(22.49 \text{ MHz}; \text{CDCl}_3)$ 213.8 (C-5), 208.8 (C-20), 84.8 (C-4) and 67.9 (C-3) (Found: C, 80.4; H, 9.7. C₂₁H₃₀O₂ requires C, 80.21; H, 9.62%).

5,5;20,20-Bis(ethylenedioxy)-4,5-secopregn-3-yne 13.-4,5-Secopregn-3-yne-5,20-dione 12 (2.00 g, 6.37 mmol), toluene-*p*-sulfonic acid (100 mg) and ethylene glycol (20 cm³) in benzene (120 cm³) were heated under reflux with water separation for 18 h. The mixture was washed with aq. 5% sodium hydrogen carbonate and saturated aq. sodium chloride, then dried and the solvents evaporated. The yellow oily residue was chromatographed on a silica column (elution with 60% diethyl ether in light petroleum) to give the title compound 13 (1.41 g, 55%); m.p. 151–153 °C; v_{max}/cm^{-1} 3268 (C=C-H); $\delta_{H}(400 \text{ MHz}; \text{CDCl}_{3})$ 0.75, 0.96 and 1.27 (each 3 H, s, 18-, 19- and 21-CH₃, respectively) and 3.94 (8 H, br m, 2 × OCH₂CH₂O); $\delta_{C}(22.49 \text{ MHz}; \text{CDCl}_{3})$ 113.7 (C-5), 111.8 (C-20), 86.4 (C-4) and 67.2 (C-3) (Found: C, 74.5; H, 9.7. C₂₅H₃₈O₄ requires C, 74.59; H, 9.51%).

5,5;20,20-Bis(ethylenedioxy)-4,5-secopregnan-4-al 14.—Disiamylborane [(Me₂CHCHMe₂)₂BH] was prepared in a 100 cm³ flask. A static pressure of nitrogen was maintained until after the oxidation. The flask was immersed in an ice-salt bath and the temperature lowered to -10 °C. A solution of 5,5;20,20bis(ethylenedioxy)-4,5-secopregn-3-yne 13 (1.207 g, 3 mmol) in THF (5 cm³) was added as rapidly as possible while maintaining the temperature below 10 $^{\circ}$ C. The cold bath was removed and the reaction mixture allowed to come to room temp., and stirred for 5 h to complete the hydroboration. The reaction mixture was cooled to 0 °C and the product was oxidised at 0 °C by the addition of hydrogen peroxide (2 cm³ 15%), maintaining the pH of the solution at approximately pH ~8 by the controlled addition of sodium hydroxide (2 mol dm^{-3} ; 1.5 cm³). After oxidation, the reaction mixture was neutralised and the organic phase was subsequently extracted into dichloromethane, dried over anhydrous sodium sulfate, and then the solvent was evaporated. An oily, colourless residue (1.15 g, 2.74 mmol) (91% crude yield) was obtained which was then chromatographed on a silica column (elution with 70% diethyl ether in light petroleum) and 5,5;20,20-bis(ethylenedioxy)-4,5-secopregnan-4-al 14 was isolated (0.66 g, 52% yield); m.p. 159–161 °C; v_{max}/cm^{-1} 2720 (OC–H) and 1700 (C=O); $\delta_{\rm H}(220 \text{ MHz}, \text{CDCl}_3) 0.76, 0.98 \text{ and } 1.20 \text{ (each } (3 \text{ H}, \text{ s}, 18\text{-}, 19\text{-})$ and 21-CH₃, respectively), 3.95 (8 H, br m, $2 \times OCH_2CH_2O$) and 9.76 (1 H, s, CHO); $\delta_{\rm C}(22.49 \, {\rm MHz}; {\rm CDCl}_3)$ 202.9 (4-CHO), 113.7 (C-5) and 111.8 (C-20) (Found: C, 71.2; H, 9.6%; M⁺, 420.2870. C₂₅H₄₀O₅ requires C, 71.39; H, 9.59%; M, 420.2876).

Wittig Product 15.—To a solution of 5,5;20,20-bis(ethylenedioxy)-4,5-secopregnan-4-al 14 (315 mg, 0.75 mmol) in dichloromethane (5 cm³) was added the reagent (methoxycarbonylmethylene)triphenylphosphorane (Ph₃P=CHCO₂Me) (1.1 equiv.) dissolved in dichloromethane. The reaction mixture was stirred at room temp. for 3 h and then refluxed for 1.5 h. Finally, it was stirred at room temp. for a further 48 h. The product was then extracted into dichloromethane and the organic extract was evaporated. The crude oily product was subsequently chromatographed (elution with 50% diethyl ether to light petroleum to give the pure Wittig product 15 (179 mg, 50%); v_{max}/cm^{-1} 3040 (C=C-H), 1720 (C=O) and 1190 (C-O); $\delta_{\rm H}(220 \text{ MHz}; \text{CDCl}_3) 0.59, 0.94 \text{ and } 1.23 \text{ (each 3 H, s, 18-, 19-)}$ and 21-CH₃, respectively), 3.71 (3 H, s, OCH₃), 3.95 (8 H, br m, $2 \times OCH_2CH_2O$, 5.83 (1 H, dt, J_{AB} 15, $J_{BX} \sim 1$, H_B) and 6.91 (1 H, dt, J_{AB} 15, $J_{AX} \sim 5$, H_A); $\delta_C(22.49 \text{ MHz}; \text{CDCl}_3)$ 175.1 (CO₂), 149.8 (C-5), 120.8 (C-4), 111.8 (C-20), 113.8 (C-5) and 51.2 (OCH₃) (Found: C, 70.4; H, 9.4. C₂₈H₄₄O₆ requires C, 70.54; H, 9.31%).

5,20-Dioxo-4-nor-3,5-secopregnan-3-oic Acid 17.—A solution of progesterone (10.00 g, 31.8 mmol) in dichloromethane–ethyl acetate (2:1; 150 cm³) was oxidised by passing ozone through the solution at -78 °C. The progress of the reaction was monitored by TLC. As soon as the progesterone had reacted (TLC analysis after 3 h) the solvents were evaporated. The residue was dissolved in glacial acetic acid and hydrogen peroxide (30%; 12.5 cm³) was added and the solution left at room temp. overnight. The solvents were then removed under reduced pressure and the residue taken up in diethyl ether and extracted with aq. sodium hydroxide (2 mol dm⁻³). This extract was acidified with concentrated hydrochloric acid and the resulting crystalline precipitate filtered off and dried. On recrystallisation from acetone–hexane the title product **17** was obtained (9.50 g, 89%); m.p. 177–179 °C; v_{max}/cm^{-1} 3354 (O–H), 1734 (C=O), 1702 (C=O) and 1293 (C–O); $\delta_{H}(400$ MHz; CDCl₃) 0.66, 1.10 and 2.11 (each 3 H, s, 18-, 19- and 21-CH₃, respectively) and 2.53 (2 H, m, 2-H); $\delta_{C}(22.49$ MHz; CDCl₃) 214.4 (C-20), 209.2 (C-5), 179.4 (C-3) and 63.2 (C-17) (Found: C, 71.7; H, 9.0. C₂₀H₃₀O₄ requires C, 71.82; H, 9.04%).

20-Acetoxy-4-oxa-3,5-secopregna-5,17-dien-3-one 18.-5,20-Dioxo-4-nor-3,5-secopregnan-3-oic acid 17 (3.00 g, 8.96 mmol) was dissolved in acetic anhydride (20 cm³) and acetyl chloride (10 cm³) and the mixture was heated under gentle reflux under an inert atmosphere in moisture-free conditions for 28 h. The solvents were then evaporated under reduced pressure at 50 °C and the brown crystalline residue was chromatographed over silica (elution with diethyl ether) giving the title product (1.44 g, 45%); m.p. 170–171 °C; v_{max}/cm⁻¹ 1751 (C=O), 1743 (C=O), 1685 (C=C), 1161 (C–O), 1734 (C=O) and 1227 (C–O); δ_H(400 MHz; CDCl₃) 0.92, 1.11 and 1.88 (each 3 H, s, 18-, 19- and 21-CH₃, respectively), 2.09 (3 H, s, acetate CH₃), 2.61–2.64 (2 H, m, 2-H) and 5.26 (1 H, dd, 6-H); $\delta_{\rm C}(22.49 \text{ MHz}; \text{CDCl}_3)$ 168.9 (ester C=O), 168.3 (C-3), 154.2 (C-5), 136.9 (C-17), 136.4 (C-20) and 105.3 (C-6) (Found: C, 73.1; H, 8.5. C₂₂H₃₀O₄ requires C, 73.71; H, 8.43%).

4-Cyanopregn-4-ene-3,20-dione 8 (4-Cyanoprogesterone).—A solution of acetonitrile (4 mmol) in THF (5 cm³) was cooled to -78 °C and BuLi (4 mmol) was added dropwise. The solution was stirred for 30 min prior to the addition of 20-acetoxy-4oxa-3,5-secopregna-5,17-dien-3-one 18 (2 mmol). After 1 h at -78 °C and 2 h at room temp., *tert*-butyl alcohol (10 cm³) containing potassium tert-butoxide (6 mmol) was added, and the solution was stirred for 16 h at room temp. The products were extracted into dichloromethane and then purified by silica chromatography eluting with 90% of diethyl ether to light petroleum. It was further recrystallised from acetone-diethyl ether (4:1) to give 4-cyanoprogesterone 8 (0.29 g, 42%); m.p. 210–212 °C (lit., ⁸ 214–217 °C); v_{max}/cm^{-1} 2233 (C=N), 1697 (C=O), 1685 (C=O) and 1586 (C=C); $\delta_{\rm H}$ (400 MHz; CDCl₃) 0.70, 1.27 and 2.14 (each 3 H, s, 18-, 19- and 21-CH₃, respectively), 3.07 (1 H, dm, J 15.3, 6-α-H); δ_C(22.49 MHz; CHCl₃) 208.8 (C-20), 192.0 (C-3), 183.4 (C-5), 113.9 (C-4) and 112.3 (CN) (Found: C, 77.6; H, 8.85; N, 4.2%; M^+ , 339.2187. $C_{22}H_{30}O_4$ requires C, 77.84; H, 8.61; N, 4.13%; M, 339.2198).

Acknowledgements

We thank the Cancer Research Campaign for generous funding of this work.

References

- 1 B. W. Metcalf, M. A. Levy and D. A. Holt, *Trends Pharm. Sci.*, 1989, **10**, 491.
- 2 T. Liang, G. H. Rasmusson and J. R. Brooks, J. Steroid Biochem., 1983, 19, 385.
- 3 T. R. Blohm, B. W. Metcalf, M. E. Laughlin, A. Sjoerdsma and G. L. Schatzman, *Biochem. Biophys. Res. Commun.*, 1980, 95, 273.
- 4 V. Petrow, G. M. Padilla, K. Kendle and A. Tantawi, J. Endocrinol., 1982, 95, 311.
- 5 M. A. Levy, M. Brandt, J. R. Heys, D. A. Holt and B. W. Metcalf, *Biochemistry*, 1990, **29**, 2815.
- 6 (a) J. Mann and B. Pietrzak, J. Chem. Soc., Perkin Trans. 1, 1983, 2681; (b) J. Mann and B. Pietrzak, J. Chem. Soc., Perkin Trans. 1,

1987, 385; (c) M. G. Rowlands, A. B. Foster, J. Mann, B. Pietrzak, J. Wilkinson and R. C. Coombes, *Steroids*, 1987, **49**, 371; (d) J. Mann and B. Pietrzak, *Tetrahedron*, 1989, **45**, 1549.

- 7 M. Haase-Held, M. Hatzis and J. Mann, J. Chem. Soc., Perkin Trans. 1, 1992, 2999.
- B. C. Huynh and S. Julia, Bull. Soc. Chim. Fr., 1971, 4396.
 E. Piers and I. F. Fleming, J. Chem. Soc., Chem. Commun., 1989, 756.
- 10 A. Eschenmoser, D. Felix and G. Ohloff, Helv. Chim. Acta, 1967, 50, 708.

Paper 3/04733D Received 6th August 1993 Accepted 15th September 1993